

M1.(a) Secondary

1

(b) Nitrogen and oxygen are very electronegative

1

Therefore, C=O and N-H are polar

1

Which results in the formation of a hydrogen bond between O and H

1

In which a lone pair of electrons on an oxygen atom is strongly attracted to the  $\delta^+H$

1

[5]

M2. (a) (i) hydrolysis

*not hydration*

1

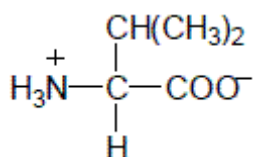
(ii) 2-aminopropanoic acid

*ignore alanine*

*QoL*

1

(iii)



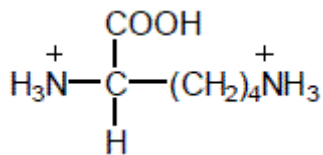
*allow  $-\text{CO}_2^-$*

*allow  $^+\text{NH}_3-$*

*don't penalize position of + on  $\text{NH}_3$*

1

(iv)



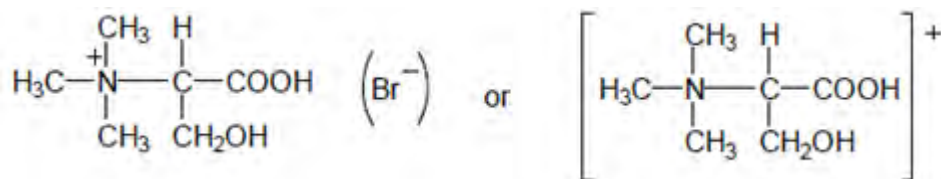
allow  $-\text{CO}_2^-$

allow  $^+\text{NH}_3-$

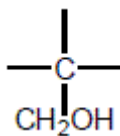
don't penalize position of + on  $\text{NH}_3$

1

(b) (i)



allow  $-\text{CO}_2\text{H}$

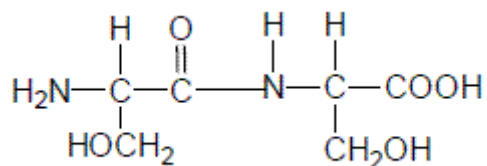


allow limit as

+ on N or outside [ ]

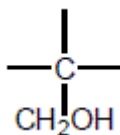
1

(ii)



allow  $-\text{CO}_2\text{H}$  allow  $-\text{CONH}-$  or  $-\text{COHN}-$

allow  $\text{NH}_2-$

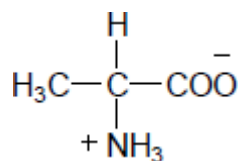


allow limit as

1

[6]

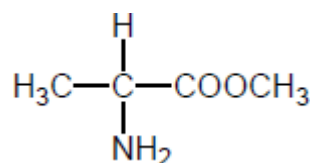
M3.(a)



Allow  $-\text{NH}_3^+$  and  $^+\text{NH}_3-$

1

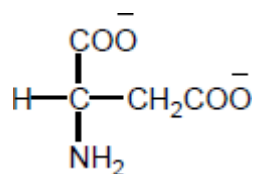
(b)



Allow protonated form, i.e.  $-\text{NH}_3^+$  or  $^+\text{NH}_3-$

1

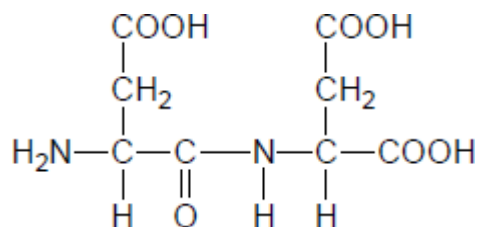
(c)



Allow  $-\text{CO}_2^-$

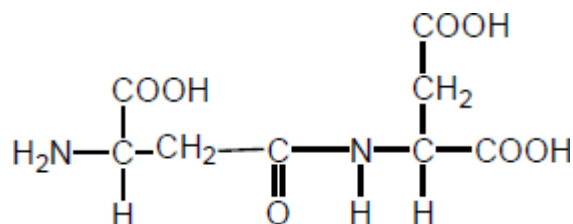
1

(d)



Allow zwitterion with any  $\text{COO}^-$

Allow use of "wrong"  $\text{COOH}$



1

[4]

M4.(a) Heating speeds up (hydrolysis / breaking of peptide bonds)

**OR** forms non-sweet (amino acids)

1

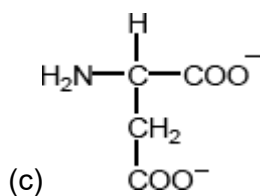
(b) (2-)aminobutanedioic acid OR

*2 not necessary but penalise other numbers at start*

(2-)aminobutane(-1,4-)dioic acid

*1,4 not necessary but penalise other numbers and 1,4 must be in correct place (QoL)*

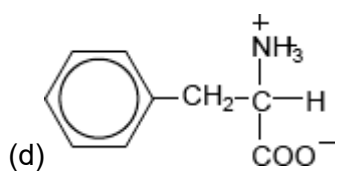
1



*allow -CO<sub>2</sub><sup>-</sup>*

*allow NH<sub>2</sub>-*

1



*allow -CO<sub>2</sub><sup>-</sup>*

*allow +NH<sub>3</sub>-*

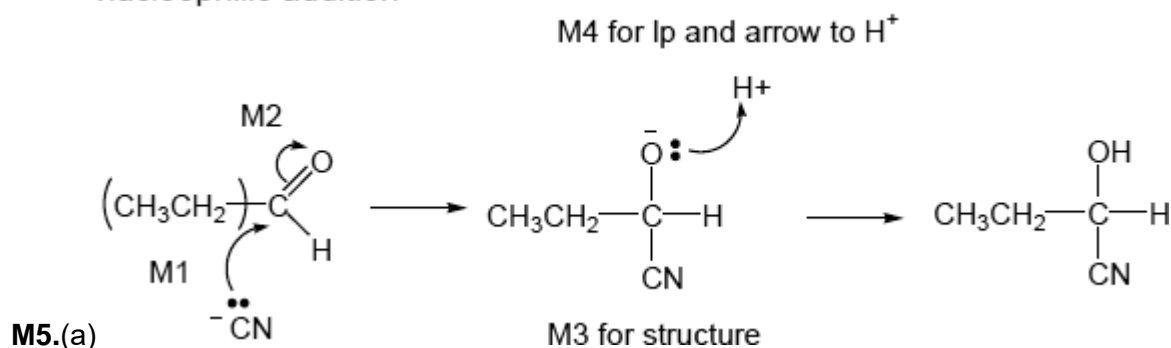
*don't penalize position of + on NH<sub>3</sub>*

1

- (e) (i) **M1** Compounds/molecules with same structural formula  
*Not just structure* 1
- M2** But with bonds/atoms/groups arranged differently in space or in 3D  
*Allow –with different spatial arrangement of atom/bond/group* 1
- Independent marks*
- (ii) (Plane) polarised light 1
- Rotated in opposite directions  
*Not bent or turned or twisted; not different directions (QoL)* 1

[8]

nucleophilic addition



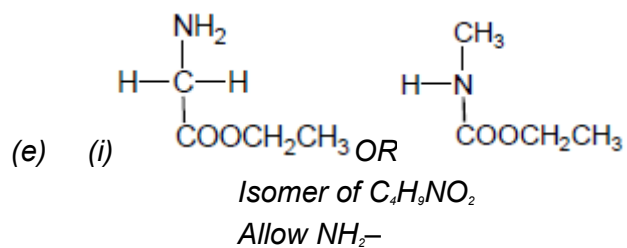
- allow :CN<sup>-</sup>
- M2 not allowed independent of M1, but
- allow M1 for correct attack on C<sup>+</sup>
- + rather than δ<sup>+</sup> on C=O loses M2
- M3 is for correct structure including minus sign but lone pair is part of M4
- Allow C<sub>2</sub>H<sub>5</sub>
- M1 and M4 for lp and curly arrow

1

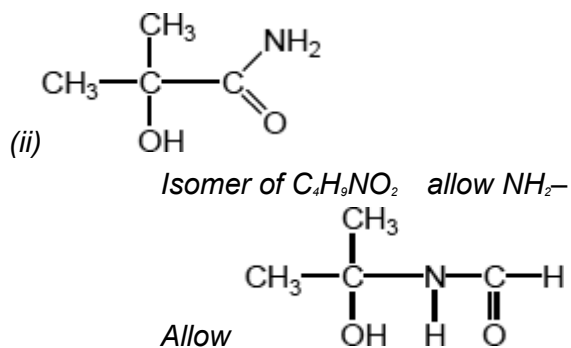
4

- (b) 2-bromobutanenitrile  
 Allow 2-bromobutane-1-nitrile  
 1
- (c) **M1** ammonia or  $\text{NH}_3$   
 Ignore temp or pressure  
 1
- M2** excess (ammonia)                      excess tied to  $\text{NH}_3$  and may score in M1 unless contradicted  
 Ignore concentrated or sealed container, Acid loses conditions mark  
 1
- M3** nucleophilic substitution  
 Allow close spelling  
 1
- (d) (i)
- $$\begin{array}{c} + \\ \text{NH}_3 \\ | \\ \text{CH}_3\text{CH}_2-\text{C}-\text{H} \\ | \\ \text{COO}^- \end{array}$$
- Allow  $\text{C}_2\text{H}_5$   
 Allow  $-\text{CO}_2^-$   
 Allow  $^+\text{NH}_3-$   
 Don't penalize position of + on  $\text{NH}_3$   
 1
- (ii) **M1** electrostatic forces between ions in X                      **QOL**  
 Allow ionic bonding.  
 1
- Marks independent
- M2** (stronger than) hydrogen bonding between  $\text{CH}_3\text{CH}_2\text{CH}(\text{OH})\text{COOH}$   
**CE** mention of molecules of X or inter molecular forces between X loses both marks

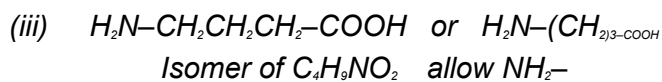
1



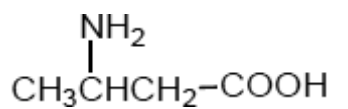
1



1



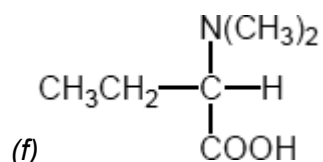
OR



*Do **not** allow -C<sub>3</sub>H<sub>6</sub>-*

*Beware – do not credit X itself*

1



*Answer has 6 carbons so **NOT** isomer of X*

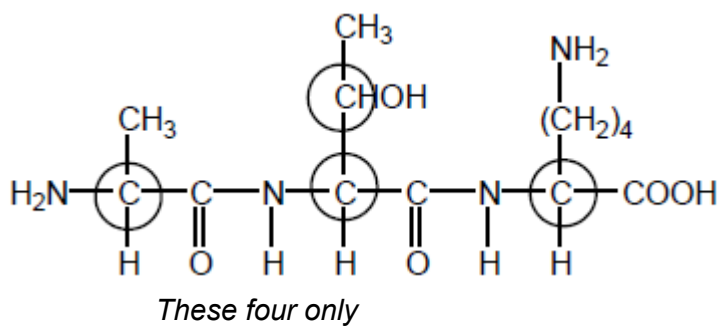
*Allow C<sub>2</sub>H<sub>5</sub>*

*Must have bond from C to N not to methyl group*

1

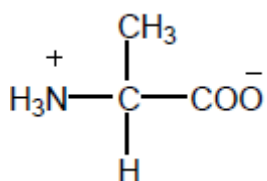
[16]

M6.(a) (i)



1

(ii)



Allow -  $\text{NH}_3^+$  and  $^+\text{NH}_3$ .

1

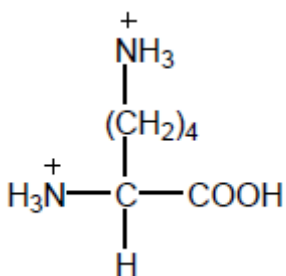
(iii) 2-amino-3-hydroxybutanoic acid  
Ignore 1 in butan-1-oic acid

Do not penalise commas or missing hyphens

Penalise other numbers

1

(iv)





Allow  $\text{-NH}_3^+$  and  $^+\text{NH}_3$ .

1

- (b) (i) **Condensation**  
Allow polyester

1

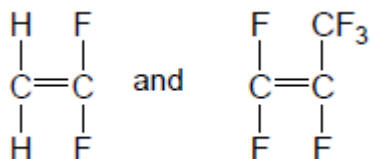
- (ii) propane-1,3-diol  
Must have e  
Allow 1,3-propanediol

1

- (c) (i) **Addition**  
Not additional

1

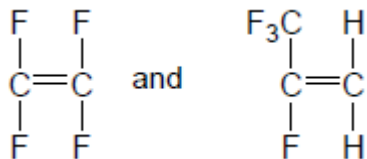
(ii)



Allow monomers drawn either way round  
Allow bond to F in  $\text{CF}_3$

1

**OR**



1 for each structure within each pair

1

- (d) c  
If wrong, CE = 0

1

*C-C or C-F bonds too strong*

*1*  
**[11]**